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* * * * * * * * * *
                     Welcome to STN International
                 Web Page for STN Seminar Schedule - N. America
NEWS
NEWS
     2 OCT 02
                 CA/CAplus enhanced with pre-1907 records from Chemisches
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NEWS 3 OCT 19
                 BEILSTEIN updated with new compounds
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                 Derwent Indian patent publication number format enhanced
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         NOV 19
                 WPIX enhanced with XML display format
NEWS 6
         NOV 30 ICSD reloaded with enhancements
NEWS 7 DEC 04 LINPADOCDB now available on STN NEWS 8 DEC 14 BEILSTEIN pricing structure to change
NEWS 9 DEC 17 USPATOLD added to additional database clusters
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NEWS 11 DEC 17 DGENE now includes more than 10 million sequences
NEWS 12 DEC 17 TOXCENTER enhanced with 2008 MeSH vocabulary in
                 MEDLINE segment
NEWS 13 DEC 17 MEDLINE and LMEDLINE updated with 2008 MeSH vocabulary
NEWS 14 DEC 17 CA/CAplus enhanced with new custom IPC display formats
NEWS 15 DEC 17
                 STN Viewer enhanced with full-text patent content
                 from USPATOLD
NEWS 16 JAN 02
                 STN pricing information for 2008 now available
NEWS 17 JAN 16
                 CAS patent coverage enhanced to include exemplified
                 prophetic substances
NEWS 18 JAN 28 USPATFULL, USPAT2, and USPATOLD enhanced with new
                 custom IPC display formats
NEWS 19 JAN 28 MARPAT searching enhanced
NEWS 20 JAN 28 USGENE now provides USPTO sequence data within 3 days
                 of publication
NEWS 21 JAN 28 TOXCENTER enhanced with reloaded MEDLINE segment
NEWS 22 JAN 28 MEDLINE and LMEDLINE reloaded with enhancements
NEWS 23 FEB 08 STN Express, Version 8.3, now available
NEWS 24 FEB 20 PCI now available as a replacement to DPCI
NEWS 25 FEB 25 IFIREF reloaded with enhancements
NEWS 26 FEB 25
                 IMSPRODUCT reloaded with enhancements
NEWS 27 FEB 29
                 WPINDEX/WPIDS/WPIX enhanced with ECLA and current
                 U.S. National Patent Classification
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NEWS EXPRESS FEBRUARY 08 CURRENT WINDOWS VERSION IS V8.3,
AND CURRENT DISCOVER FILE IS DATED 20 FEBRUARY 2008

NEWS HOURS STN Operating Hours Plus Help Desk Availability
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=> (acridine or dye or ethidium) (15A) (DNA or nucleic or RNA) (metal or toxicant or chromium or mercury or pollutant) MISSING OPERATOR RNA) (METAL

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

```
=> (acridine or dye or ethidium) (15A) (DNA or nucleic or RNA) (15A) (metal or toxicant
or chromium or mercury or pollutant)
             4 FILE AGRICOLA
L1
L2
             9 FILE BIOTECHNO
L3
             2 FILE CONFSCI
L4
             1 FILE HEALSAFE
L5
            21 FILE LIFESCI
L6
            34 FILE PASCAL
TOTAL FOR ALL FILES
            71 (ACRIDINE OR DYE OR ETHIDIUM) (15A) (DNA OR NUCLEIC OR RNA) (15A) (M
               ETAL OR TOXICANT OR CHROMIUM OR MERCURY OR POLLUTANT)
=> 17 and (displacement or competition or compete)
             0 FILE AGRICOLA
L8
             0 FILE BIOTECHNO
L9
L10
             0 FILE CONFSCI
             O FILE HEALSAFE
L11
L12
             3 FILE LIFESCI
L13
             1 FILE PASCAL
TOTAL FOR ALL FILES
L14
             4 L7 AND (DISPLACEMENT OR COMPETITION OR COMPETE)
=> d l14 ibib abs total
L14 ANSWER 1 OF 4 LIFESCI
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ACCESSION NUMBER:
                    2007:140853 LIFESCI
TITLE:
                    Synthesis, characterization and DNA-binding properties of
                    zinc(II) and nickel(II) Schiff base complexes
                    Qi, Gao-fei; Yang, Zheng-yin*; Wang, Bao-dui
AUTHOR:
CORPORATE SOURCE:
                    Lanzhou University, Lanzhou, 730000, P.R. China; E-mail:
                    yangzy@lzu.edu.cn
SOURCE:
                    Transistion Metal Chemistry [Transistion Met. Chem.],
                    (20070300) vol. 32, no. 2, pp. 233-239.
                    ISSN: 0340-4285.
DOCUMENT TYPE:
                    Journal
FILE SEGMENT:
                    Ν
LANGUAGE:
                    English
                   English
SUMMARY LANGUAGE:
     A new ligand, 2-carboxybenzaldehyde-(4'-hydroxy)benzoylhydrazone(H
     sub(2)L) and its In super(II) and Ni super(II) complexes have been
     synthesized and characterized on the basis of elemental analyses, molar
     conductivities, super(1) H-NMR, IR spectra and thermal analyses. In
     addition, DNA-binding properties of these two metal
     complexes were investigated using spectrometric titrations,
     ethidium bromide displacement experiments, and viscosity
     measurements. The results show that the two complexes, especially the Ni
     super(II) complex, strongly bind with calf-thymus DNA, presumably via an
     intercalation mechanism. The intrinsic binding constants of the Zn
     super(II) and Ni super(II) complexes with DNA are 2.46 x 10 super(5) and
     7.94 \times 10 \text{ super}(5) \text{M super}(-1), \text{ respectively.}
L14 ANSWER 2 OF 4 LIFESCI
                               COPYRIGHT 2008 CSA on STN
                    93:6708 LIFESCI
ACCESSION NUMBER:
TITLE:
                    Fluorescence studies of dye displacement
                    from DNA by chromium(III) complexes:
                    Evidence for cation induced DNA condensations.
                    Kortenkamp, A.; O'Brien, P.; Beyersmann, D.
AUTHOR:
```

CORPORATE SOURCE: Sch. Pharm., Univ. London, 29/39 Brunswick Square, London

WC1N 1AX, UK

SOURCE: TOXICOL. ENVIRON. CHEM., (1992) vol. 35, no. 3-4, pp.

149-159.

DOCUMENT TYPE: Journal FILE SEGMENT: X; N LANGUAGE: English SUMMARY LANGUAGE: English

The interactions of 10 different chromium(III) complexes with isolated calf thymus DNA have been analysed by studying the electronic and fluorescence spectra of intercalated ethidium bromide. Triply charged cationic complexes including: (Cr(urea) sub(6))Cl sub(3).3H sub(2)O, (Cr(1,10-phenanthroline) sub(3)) (ClO sub(4)) sub(3).2Hsub(2)0, (Cr(2,2'-bipyridyl) sub(3)) (ClO sub(4)) sub(3).2H sub(2)0, (Cr (ethylendiamine) sub(3)) Cl sub(3) 3.5H sub(2)O and (Cr(NH sub(3)) sub(6)) (NO sub(3)) sub(3) displaced the dye from DNA. Similar effects were observed in experiments using the non-intercalating dye bisbenzimidazole ("Hoechst 33258"). Singly charged cationic, anionic and uncharged chromium(III) complexes such as: cis-(Cr(1,10phenanthroline) sub(2)Cl sub(2)) Cl.2H sub(2)O, cis-(Cr(2,2'-bipyridyl) sub(2)Cl sub(2)) Cl.2H sub(2)O, (Cr(glutathione) sub(2)) Na sub(2), (Cr(cysteine) sub(2))Na.2H sub(2)O and (Cr(glycine) sub(3)) were unable to displace both ethidium-bromide and bisbenzimidazole from DNA. There was no evidence for the formation of co-ordinate bonds between chromium(III) and DNA for any of the above complexes. The charge and type of ligand are important in controlling the interaction of chromium(III) with isolated DNA in vitro.

L14 ANSWER 3 OF 4 LIFESCI COPYRIGHT 2008 CSA on STN

ACCESSION NUMBER: 81:10387 LIFESCI

TITLE: Metal Mutagens and Carcinogens Effectively

Displace Acridine Orange From DNA as Measured by Fluorescence Polarization.

AUTHOR: Richardson, C.L.; Verna, J.; Schulman, G.; Shipp, K.;

Grant, A.D.

CORPORATE SOURCE: Meloy Lab., 6715 Electronic Dr., Springfield, VA 22151, USA

ENVIRON. MUTAGENESIS., (1981) vol. 3, no. 5, pp. 545-553.

DOCUMENT TYPE: Journal FILE SEGMENT: N; G; X LANGUAGE: English SUMMARY LANGUAGE: English

SOURCE:

Displacement of 50% of the acridine orange is obtained with less than 0.5 mM concentrations of lead, manganese, cobalt, zinc, cadmium, nickel, iron, copper, and cis-platinum. In contrast, greater than 80 mM concentrations of lithium, sodium, and potassium are required to displace an equivalent amount of acridine orange from calf thymus DNA. Although cis-platinum shows the best DNA reactivity in this assay, the interaction between this metal and DNA does not occur immediately, as it does for the other metals tested. These results indicate that the acridine orange displacement assay provides a relative measure of the interaction of metals with DNA, and this DNA reactivity shows a positive correlation with mutagenic/carcinogenic potential.

L14 ANSWER 4 OF 4 PASCAL COPYRIGHT 2008 INIST-CNRS. ALL RIGHTS RESERVED. on

ACCESSION NUMBER: 2006-0532862 PASCAL

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TITLE (IN ENGLISH): Improved curve fitting procedures to determine

equilibrium binding constants

AUTHOR: STOOTMAN Frank H.; FISHER Dianne M.; RODGER Alison;

ALDRICH-WRIGHT Janice R.

CORPORATE SOURCE: University of Western Sydney, PO Box 1797, Penrith

South DC, NSW 1797, Australia; The University of Sydney, Centre for Heavy Metals Research, School of Chemistry, The University of Sydney, NSW2006,

Australia; The University of Warwick, Coventry, CV4

7AL, United Kingdom

SOURCE: Analyst : (London. 1877. Print), (2006), 131(10),

1145-1151, 24 refs.

ISSN: 0003-2654 CODEN: ANALAO

DOCUMENT TYPE: Journal
BIBLIOGRAPHIC LEVEL: Analytic
COUNTRY: United Kingdom

LANGUAGE: English

AVAILABILITY: INIST-1036, 354000158753140120

AN 2006-0532862 PASCAL

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AB For ligand-biomacromolecule titration experiments it has been traditional practice to extract parameters such as the equilibrium binding constant K and the number of bases per ligand binding site n with relatively labour intensive methods, usually based on single wavelength data, such as the difference method by Rodger and Norden coupled together with a Scatchard plot. Presented in this paper are both the theory and a least squares fitting method to derive parameters such as K and n more directly from all spectral non-linear experimental data. Both the case of non competitive binding of a metal complex ligand to DNA and the case of displacement by a metal complex ligand of an ethidium marker attached to the DNA are considered. This work may be applied directly to reduce experimental data produced by a spectropolarimeter (for circular or linear dichroism) or a spectrophotometer (for fluorescence or UV-Vis spectroscopy).